

# The ionization, excitation and thermal state of flows in the ISM

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# 1. Gasdynamic equations (3D, Cartesian)

continuity equation  $\rightarrow$

$$\frac{\partial \rho}{\partial t} + \frac{\partial}{\partial x_j} \rho u_j = 0, \quad (1)$$

momentum equation (3 components)  $\rightarrow$

$$\frac{\partial \rho u_i}{\partial t} + \frac{\partial}{\partial x_j} (\rho u_i u_j + P \delta_{ij}) = 0, \quad (2)$$

energy equation  $\rightarrow$

$$\frac{\partial E}{\partial t} + \frac{\partial}{\partial x_j} [u_j (E + P)] = G - L, \quad (3)$$

with  $E = \rho u_j u_j / 2 + P / (\gamma - 1)$  ( $\gamma$  is the specific heat ratio,  $= 5/3$  for our discussion of a non-relativistic, atomic/ionic gas).

Heating and cooling: contributions from many species

$$G = \sum_{\alpha} G_{\alpha}; \quad L = \sum_{\alpha} L_{\alpha}, \quad (4)$$

Continuity equations for all of the species:

$$\frac{\partial n_a}{\partial t} + \frac{\partial}{\partial x_j} n_a u_j = S_a, \quad (5)$$

Alternative, transport form:

$$\frac{\partial y_a}{\partial t} + u_j \frac{\partial}{\partial x_j} y_a = \frac{S_a}{n}, \quad (6)$$

where  $y_a = n_a/n$  is the fractional abundance of the species  $a$  (with  $n = \rho/m$ ,  $m$  being the average mass per atom or ion).

## Rate equations for a constant density gas

$$\frac{dn_{a,z}}{dt} = S_{a,z}^c + S_{a,z}^{ph}, \quad (7)$$

with the collisional ( $S^c$ ) and photoionisation ( $S^{ph}$ ) source terms being given by

$$S_{a,z}^c = n_e [n_{a,z-1}c_{a,z-1} + n_{a,z+1}\alpha_{a,z+1} - n_{a,z}(c_{a,z} + \alpha_{a,z})], \quad (8)$$

$$S_{a,z}^{ph} = n_{a,z-1}\phi_{a,z-1} - n_{a,z}\phi_{a,z}. \quad (9)$$

## The electron density

$$n_e = \sum_a \sum_z z n_{a,z}. \quad (10)$$

→ Charge exchange reactions

## Collisional ionisation and radiative+dielectronic recombination coefficients:

$$\alpha(T), c(T) = \int_0^{\infty} f(v, T) \sigma_v v dv, \quad (11)$$

### Arrhenius:

It is common to give analytic fits to these coefficients in the “Arrhenius interpolation” form :

$$r(T) = b_1 T^{b_2} e^{b_3/T}. \quad (12)$$

### Aldrovandi & Pequignot (1973, 1976)

$$r(T) = b_1 \left( \frac{T}{10^4} \right)^{-b_2} + b_3 T^{-3/2} \exp(-b_4/T) [1 + b_5 \exp(-b_6/T)], \quad (13)$$

for recombination coefficients.

Table 1: Ionisation, recombination and charge exchange coefficients

reaction	coefficients <sup>a</sup>
$e + \text{HI} \rightarrow 2e + \text{HII}$	1: $5.83 \times 10^{-11}$ , 0.5, -157800
$e + \text{HII} \rightarrow \text{HI}$	1: $3.69 \times 10^{-10}$ , -0.79, 0
$e + \text{HeI} \rightarrow 2e + \text{HeII}$	1: $2.707 \times 10^{-11}$ , 0.5, -285400
$e + \text{HeII} \rightarrow \text{HeI}$	2: $4.3 \times 10^{-13}$ , 0.672, 0.0019, $4.7 \times 10^5$ , 0.3, 94000
$e + \text{HeII} \rightarrow 2e + \text{HeIII}$	1: $5.711 \times 10^{-12}$ , 0.5, -631000
$e + \text{HeIII} \rightarrow \text{HeII}$	1: $2.21 \times 10^{-9}$ , -0.79, 0
$e + \text{CII} \rightarrow 2e + \text{CIII}$	1: $3.93 \times 10^{-11}$ , 0.5, -283000
$e + \text{CIII} \rightarrow \text{CII}$	2: $3.2 \times 10^{-12}$ , 0.770, 0.038, $9.1 \times 10^4$ , 2.0, $3.7 \times 10^5$
$e + \text{CIII} \rightarrow 2e + \text{CIV}$	1: $2.04 \times 10^{-11}$ , 0.5, -555600
$e + \text{CIV} \rightarrow \text{CIII}$	2: $2.3 \times 10^{-12}$ , 0.645, $7.03 \times 10^{-3}$ , $1.5 \times 10^5$ , 0.5, $2.3 \times 10^5$
$e + \text{NI} \rightarrow 2e + \text{NII}$	1: $6.18 \times 10^{-11}$ , 0.5, -168200
$e + \text{NII} \rightarrow \text{NI}$	2: $1.5 \times 10^{-12}$ , 0.693, 0.0031 $2.9 \times 10^5$ , 0.6, $1.7 \times 10^5$
$e + \text{NII} \rightarrow 2e + \text{NIII}$	1: $4.21 \times 10^{-11}$ , 0.5, -343360
$e + \text{NIII} \rightarrow \text{NII}$	2: $4.4 \times 10^{-12}$ , 0.675, 0.0075 $2.6 \times 10^5$ , 0.7, $4.5 \times 10^5$
$e + \text{OI} \rightarrow 2e + \text{OII}$	1: $1.054 \times 10^{-10}$ , 0.5, -157800
$e + \text{OII} \rightarrow \text{OI}$	2: $2.0 \times 10^{-12}$ , 0.646, 0.0014 $1.7 \times 10^5$ , 3.3, $5.8 \times 10^4$
$e + \text{OII} \rightarrow 2e + \text{OIII}$	1: $3.53 \times 10^{-11}$ , 0.5, -407200
$e + \text{OIII} \rightarrow \text{OII}$	2: $3.1 \times 10^{-13}$ , 0.678, 0.0014 $1.7 \times 10^5$ , 2.5, $1.3 \times 10^5$
$e + \text{OIII} \rightarrow 2e + \text{OIV}$	1: $1.656 \times 10^{-11}$ , 0.5, -636900
$e + \text{OIV} \rightarrow \text{OIII}$	2: $5.1 \times 10^{-12}$ , 0.666, 0.0028 $1.8 \times 10^5$ , 6.0, 91000
$e + \text{SII} \rightarrow 2e + \text{SIII}$	1: $7.12 \times 10^{-11}$ , 0.5, -271440
$e + \text{SIII} \rightarrow \text{SII}$	2: $1.8 \times 10^{-12}$ , 0.686, 0.0049 $1.2 \times 10^5$ , 2.5, 88000
$\text{HI} + \text{NII} \rightarrow \text{HII} + \text{NI}$	1: $1.1 \times 10^{-12}$ , 0, 0
$\text{HII} + \text{NI} \rightarrow \text{HI} + \text{NII}$	1: $4.95 \times 10^{-12}$ , 0, -10440
$\text{HI} + \text{OII} \rightarrow \text{HII} + \text{OI}$	1: $2.0 \times 10^{-9}$ , 0, 0
$\text{HII} + \text{OI} \rightarrow \text{HI} + \text{OII}$	1: $1.778 \times 10^{-9}$ , 0, -220

<sup>a</sup>The interpolation formulae are of the form "1:" Arrhenius, or "2:" Aldrovandi & Péquignot (1973), see equations (12) and (13)

## Photoionization rates:

$$\phi_{a,z} = \int_{\nu_{a,z}}^{\infty} \frac{4\pi J_{\nu}}{h\nu} \sigma_{a,z}(\nu) d\nu, \quad (14)$$

## Radiative transfer equation:

$$\frac{dI_{\nu}}{dl} = j_{\nu} - \kappa_{\nu} I_{\nu} \quad (15)$$

## Average intensity:

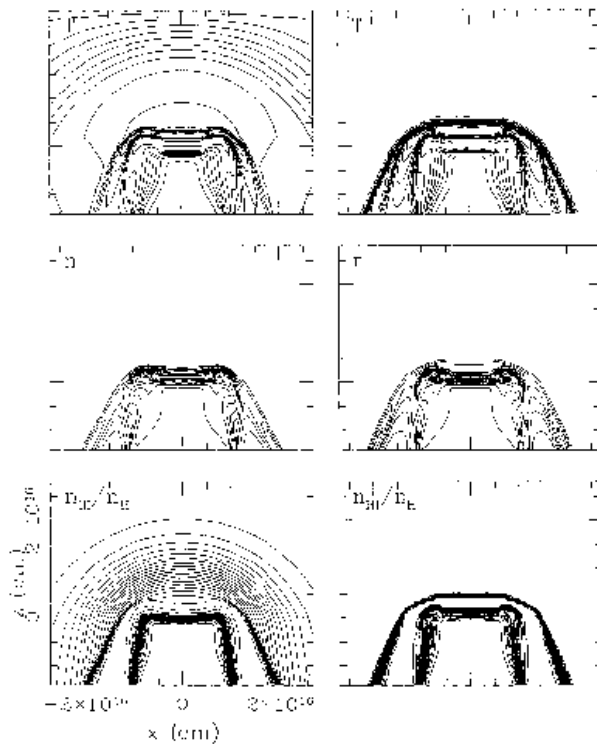
$$J_{\nu} = \frac{1}{4\pi} \oint I_{\nu} d\Omega \quad (16)$$



→ Included in 1D (stationary and time-dependent) shock models

→ Two papers in which the “diffuse” ionisation field is included

→ Of order 20 papers in which an external ionising field is included (flows in HII regions)



## Standard calculations:

- S and C are assumed to be at least singly ionised
- Photoionisation (and associated heating) neglected

## Coronal ionisation equilibrium (S=0):

$$n_{\alpha,z} c_{\alpha,z} = n_{\alpha,z+1} \alpha_{\alpha,z+1}. \quad (17)$$

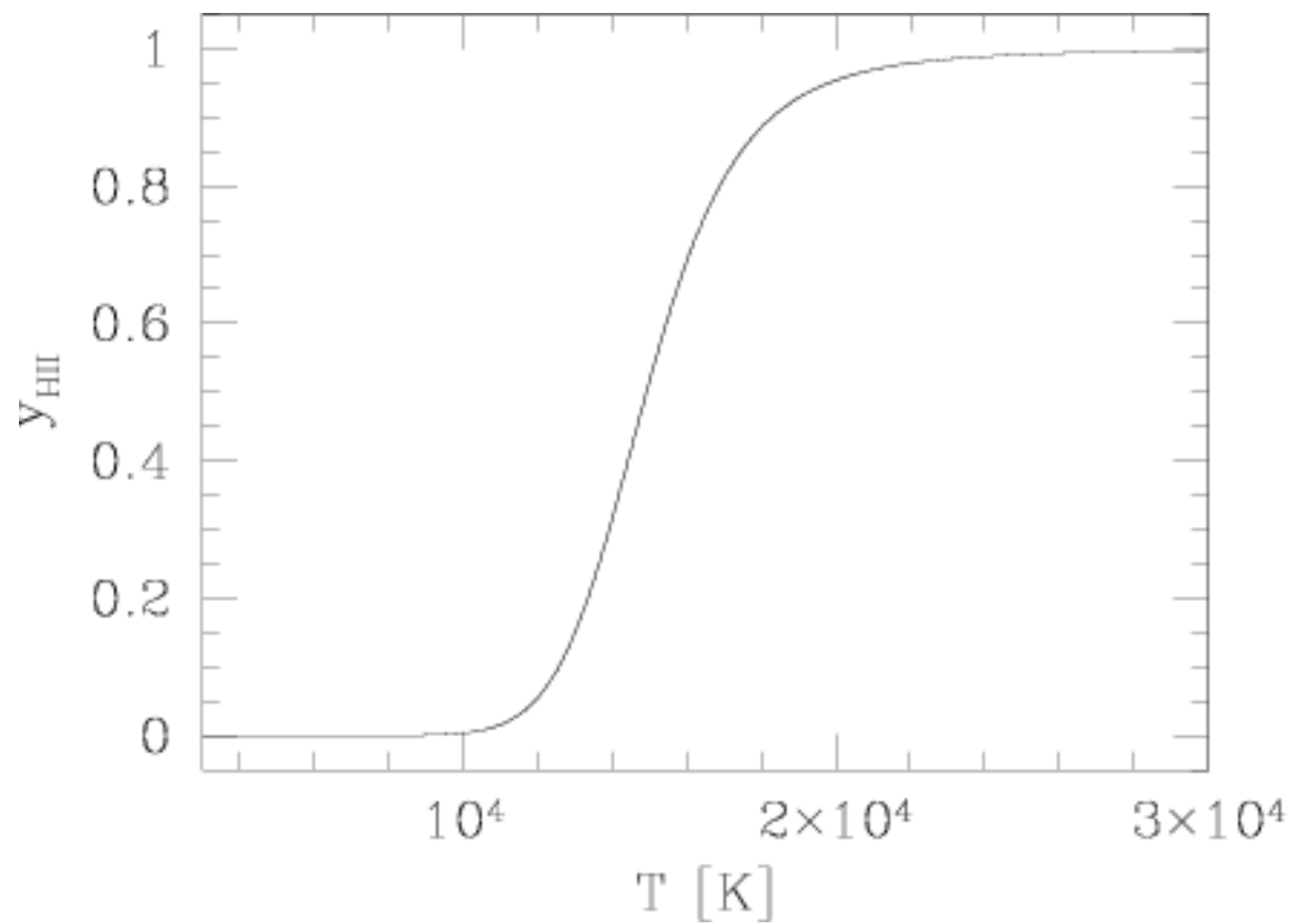
## Equilibrium for H:

$$n_{HI} c(T) = n_{HII} \alpha(T). \quad (18)$$

We can combine this equation with  $n_H = n_{HI} + n_{HII}$  to obtain

$$y_{HII} = \frac{n_{HII}}{n_H} = \frac{1}{1 + \alpha(T)/c(T)}, \quad (19)$$

where the Arrhenius interpolations for the coefficients are  $\alpha(T) = 3.69 \times 10^{-10} T^{-0.79}$  and  $c(T) = 5.83 \times 10^{-11} T^{0.5} e^{-157800/T}$



# The cooling function:

## Recombination and free-free cooling:

$$L_{ff}(HII) = n_e n_{HII} \beta_{ff}(T), \quad (20)$$

where the interpolation formula

$$\beta_{ff}(z, T) = 1.846 \times 10^{-27} z^2 T^{1/2}, \quad (21)$$

$$L_{rec}(HII) = n_e n_{HII} \beta_{rec}(T), \quad (22)$$

where the interpolation formula

$$\beta_{rec}(t) = 1.133 \times 10^{-24} t^{-1/2} \left( -0.0713 + 0.5 \ln t + 0.640 t^{-1/3} \right) \quad (23)$$

with  $t = 157890/T$  (see Seaton 1959).

$$\beta_{rec}(z, T) = z \beta_{rec}(1, T/z^2), \quad (24)$$

## Collisional ionisation:

$$L_{\alpha,z}^{\text{ion}} = n_e n_{\alpha,z} C_{\alpha,z}(T) \chi_{\alpha,z}, \quad (25)$$

## Collisionally excited lines:

$l=1, \dots, N$  : excited levels

$$n_{\alpha, z} = \sum_{l=1}^N n_l. \quad (26)$$

Energy loss  $\rightarrow$

$$L_{\alpha, z}^{\text{col}} = \sum_{l=1}^N n_l \sum_{m < l} A_{l, m} h\nu_{l, m}, \quad (27)$$

Populations of the excited levels →

$$\sum_{m>l} n_m A_{m,l} + n_e \sum_{m \neq l} n_m q_{m,l}(T) = n_l \left[ \sum_{m<l} A_{m,l} + n_e \sum_{m \neq l} q_{l,m}(T) \right], \quad (28)$$

$$n_{a,z} = \sum_{l=1}^N n_l. \quad (26)$$

→ Linear system of equations for the populations of the N levels



Collisional de-excitation coefficient →

$$q_{m,j}(T) = \frac{8.629 \times 10^{-6} \Omega_{m,j}(T)}{T^{1/2} g_m}, \quad (29)$$

Collisional excitation coefficient →

$$q_{l,m}(T) = \frac{g_m}{g_l} e^{-h\nu_{m,l}/kT} q_{m,l}(T). \quad (30)$$

Collision strengths → Chianti database

A coefficients

## The 3-level atom:

$$n_1 + n_2 + n_3 = n_{a,z}. \quad (31)$$

Now, for  $l = 1$ , from equation (28) we obtain

$$n_1 [-n_e(q_{12} + q_{13})] + n_2(A_{21} + n_e q_{21}) + n_3(A_{31} + n_e q_{31}) = 0, \quad (32)$$

and for  $l = 2$ , we obtain :

$$n_1(n_e q_{12}) + n_2[-A_{21} - n_e(q_{23} + q_{21})] + n_3(A_{32} + n_e q_{32}) = 0. \quad (33)$$

→ System of 3 linear equations

## 2-level atom:

Equation (26) takes the form

$$n_{a,z} = n_1 + n_2, \quad (34)$$

and equation (28) takes the form

$$n_1 n_c q_{12} = n_2 (n_c q_{21} + A_{21}). \quad (35)$$

Solution  $\rightarrow$

$$n_2 = \frac{n_{a,z}}{(g_1/g_2)e^{E_{21}/kT} + 1 + n_c/n_c}, \quad (36)$$

where the critical density is defined as  $n_c \equiv A_{21}/q_{12}$ .

## Low and high density regimes:

for  $n_e \ll n_c$  (the “low density regime”), we have

$$n_2 = \frac{n_{a,z} n_c q_{12}}{A_{21}}, \quad (37)$$

and for  $n_e \gg n_c$  (the “high density regime”), we have

$$n_2 = \frac{n_{a,z} g_2 e^{-E_{21}/kT}}{g_1 + g_2 e^{-E_{21}/kT}}, \quad (38)$$

## Cooling: general, low and high density regime

$$L_{21} = n_2 A_{21} h\nu_{21}, \quad (39)$$

which for the low density regime then takes the form

$$L_{21} = n_a n_e q_{12} h\nu_{21}, \quad (40)$$

and for the high density regime becomes

$$L_{21} = n_2 (L T E) A_{21} h\nu_{21}, \quad (41)$$

## Recipes for calculating the cooling function

2 most simple approaches:

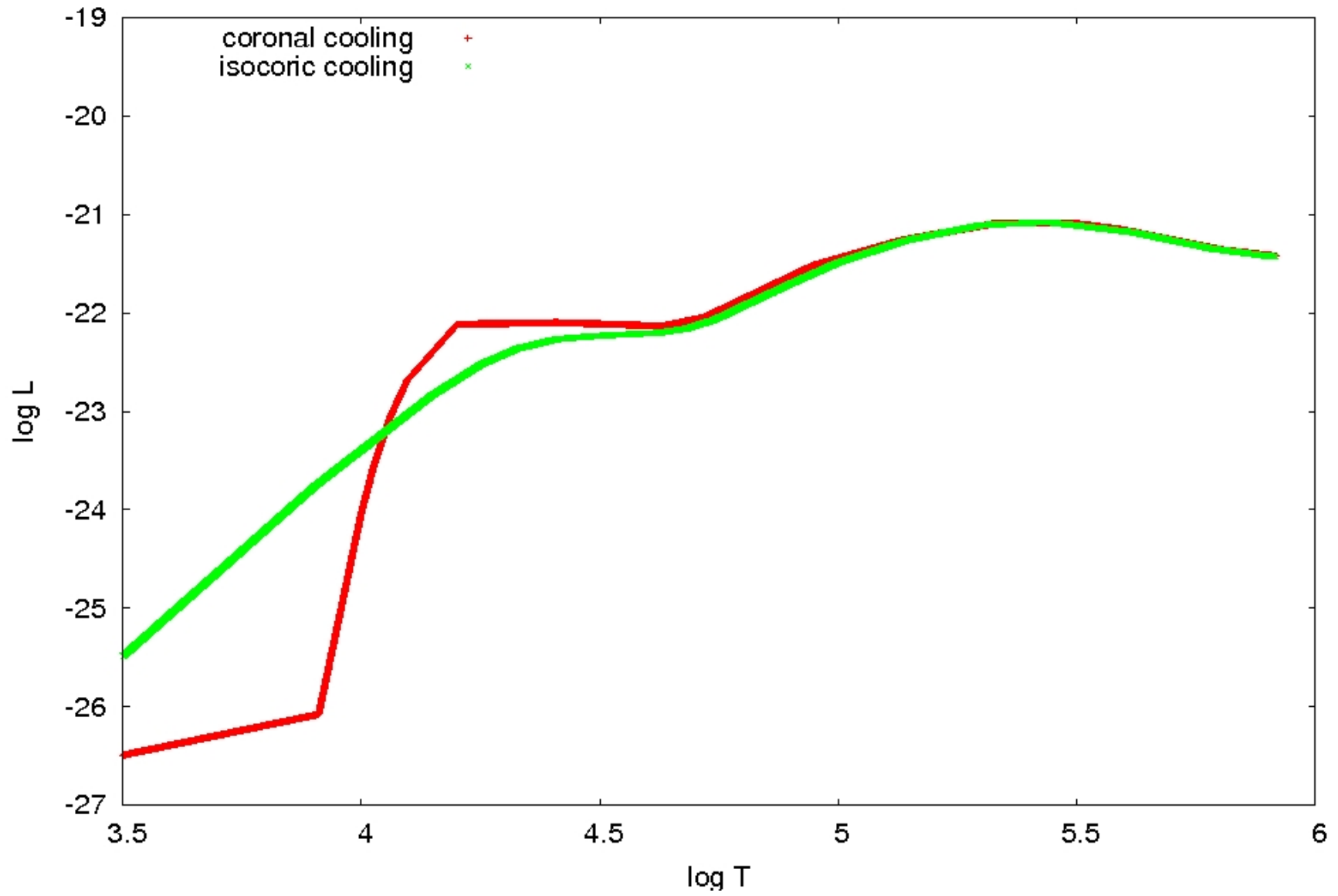
→ Coronal ionisation cooling function

→ Isochoric cooling function

$$\frac{3}{2} \frac{d}{dt} (n + n_e) kT = -L, \quad (42)$$

Both tabulated as a function of T (for n=1)

Cooling functions



→ Tabulations of cooling functions for many ions  
(as a function of  $n$  and  $T$ )

→ Fits to the low density regime cooling (as a function of  
 $T$ , for  $n=1$ )

$$\log_{10} \left( \frac{L_{OI}}{n_e n_{OI}} \right) = -45.0 + 1.23t_1 + 0.5t_1^{10} + 1.2t_2 + 1.2 [\max(t_2, 0)], \quad (43)$$

$$\log_{10} \left( \frac{L_{OII}}{n_e n_{OII}} \right) = -47.3 + 7.9t_3 + 1.9 \frac{t_4}{|t_4|^{0.5}}, \quad (44)$$

with  $t_1 = 1 - 100/T$ ,  $t_2 = 1 - 10^4/T$ ,  $t_3 = 1 - 2000/T$ ,  $t_4 = 1 - 5 \times 10^4/T$ .



## Single-species non-equilibrium cooling:

- Single continuity/rate equation for HI (or HII)
- Assume that the OI/II ionisation follows HI/II
- Consider only HI collisional ionisation + OI/OII collisionally excited line cooling
- Has to include a switch to other cooling function above 20000 K

## Cooling:

- coronal, isochoric
- single species non-equilibrium
- many species non-equilibrium

# Calculation of the emitted spectrum

the emission coefficient →

the total line emission coefficient can be computed as

$$j_{21}^{(L)} = \frac{n_2 A_{21} h \nu_{21}}{4\pi}. \quad (45)$$

The emission coefficient as a function of frequency  $\nu$  is

$$j_{21}(\nu) = \frac{j_{21}^{(L)}}{\sqrt{\pi} \Delta \nu_D} e^{-(\nu - \nu_{21})^2 / \Delta \nu_D^2}, \quad (46)$$

where  $\Delta \nu_D = \nu_{21} v_T$  with  $v_T = \sqrt{2kT/m_a}$  ( $m_a$  being the mass of the element which gives rise to the line).

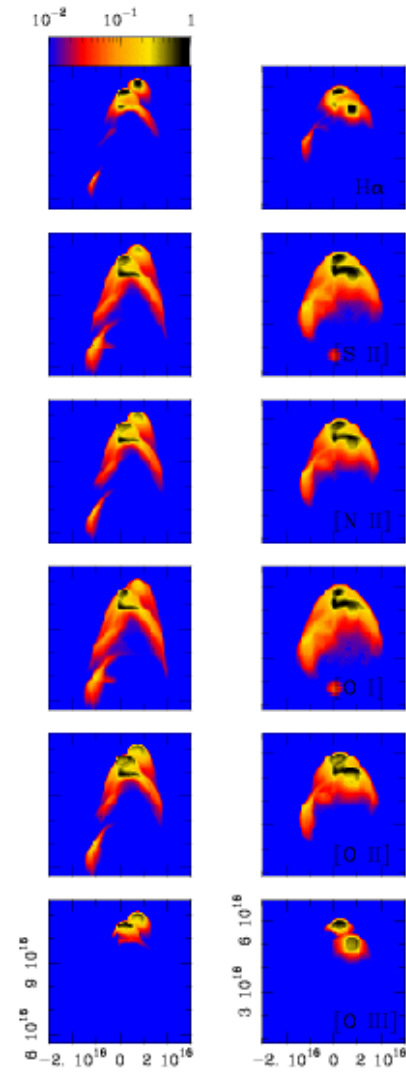
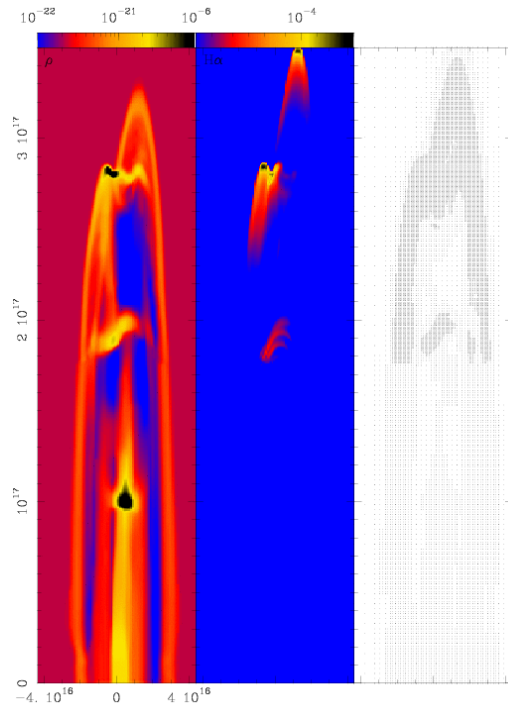
as a function of radial velocity:

$$j_{21}(v_r) = \frac{j_{21}^{(L)}}{\sqrt{\pi}v_T} e^{-v_r^2/v_T^2} . \quad (47)$$

predicted emission line

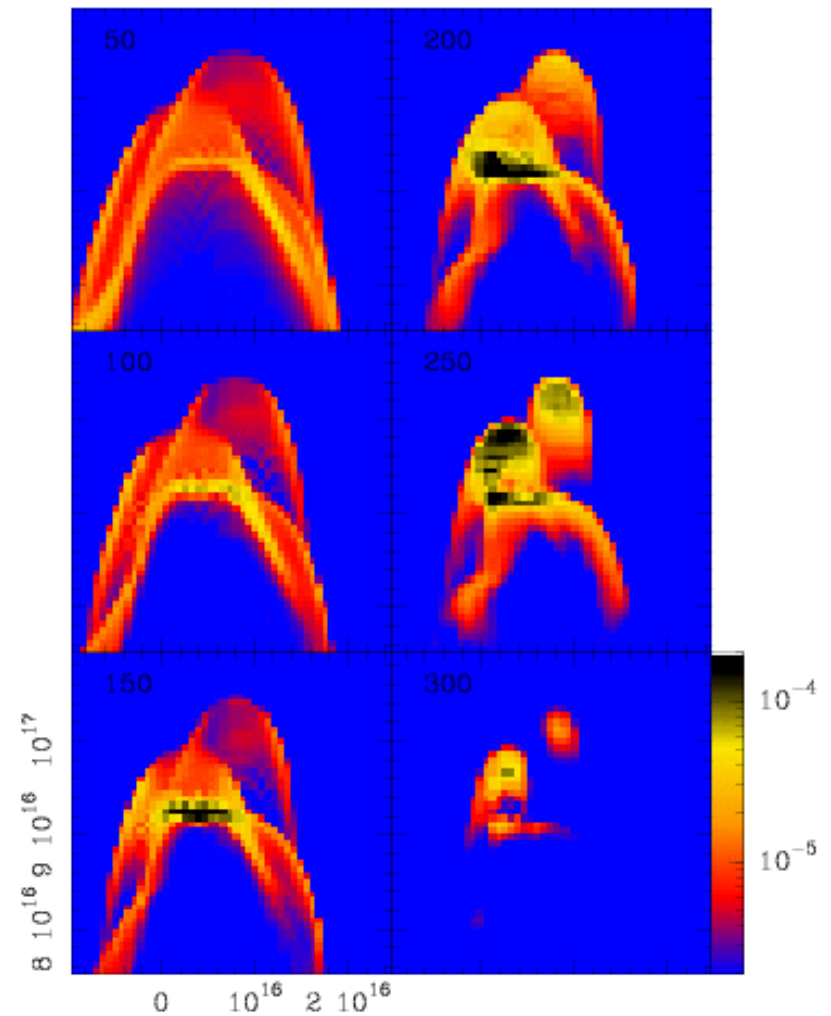
intensity map:

$$I_{21}(x, y) = \int j^{(L)}_{21}(x, y, z) dz,$$

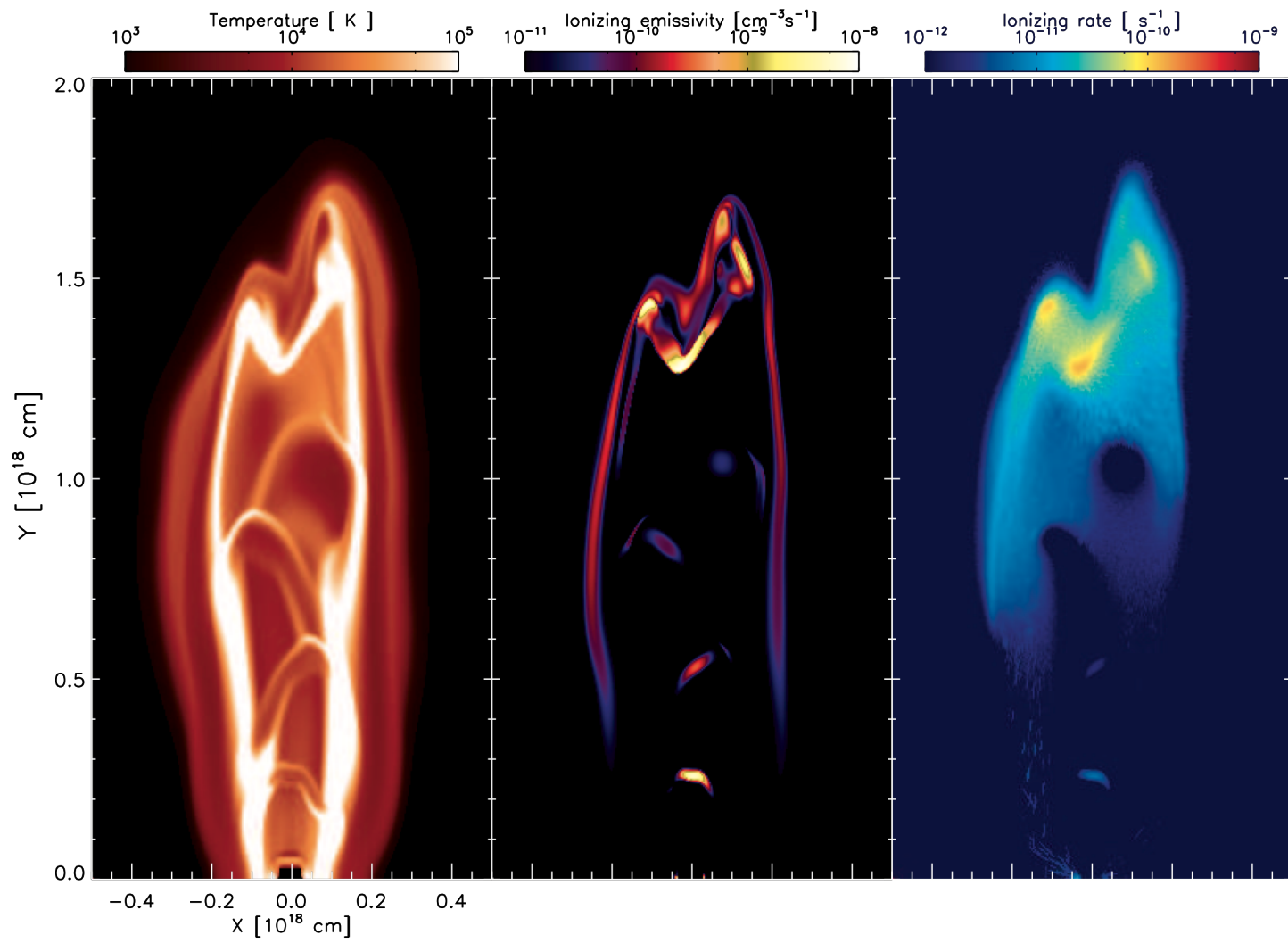


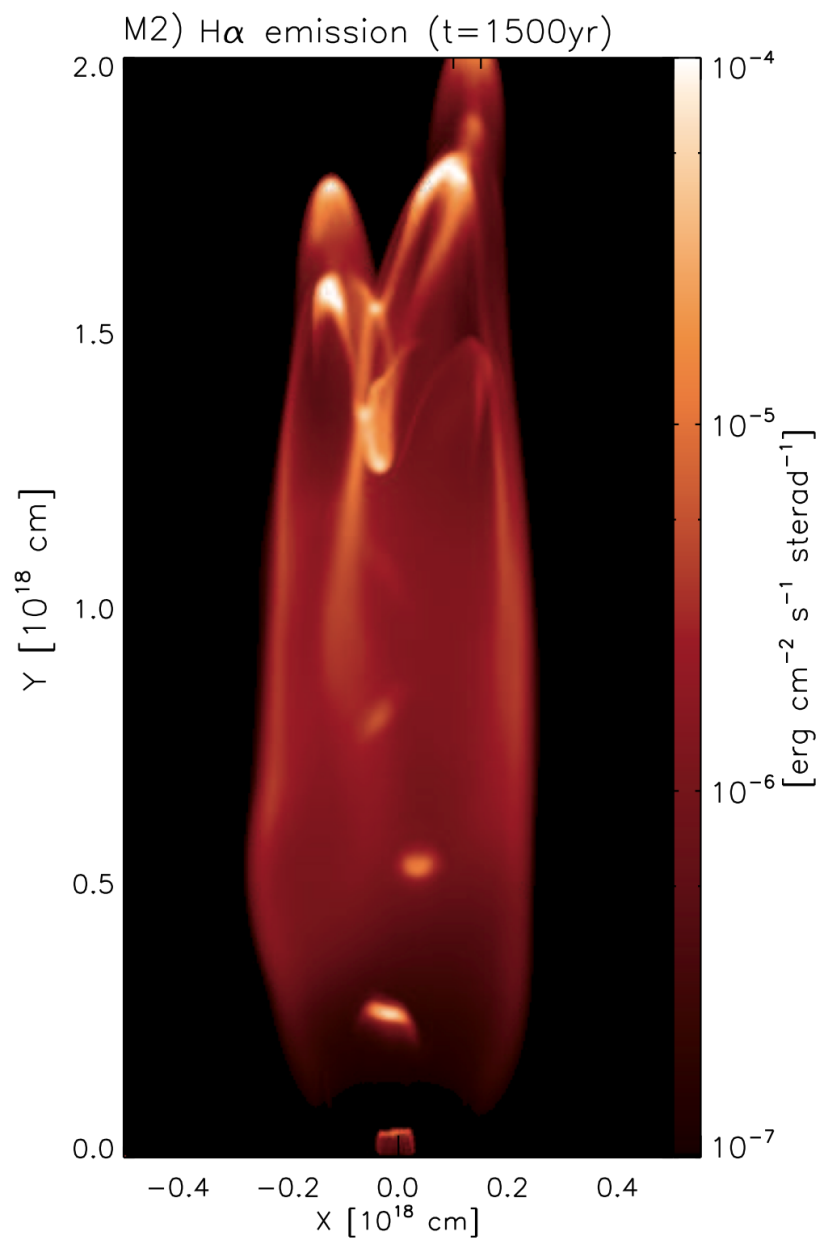
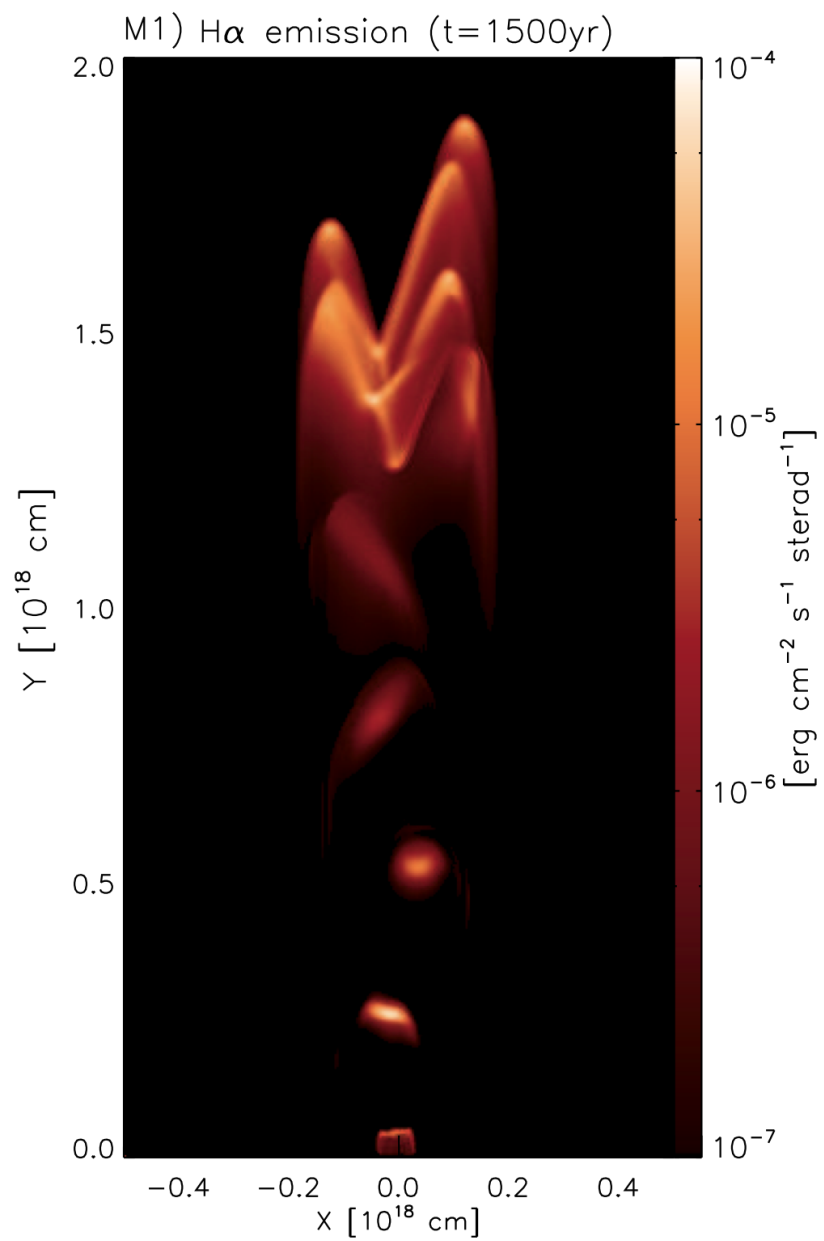
# predicted velocity channel maps

$$I_{21}(v_r, x, y) = \int j_{21}(v_r - v^{(f)}_r, x, y, z) dz, \quad (49)$$



M2) t=1500yr







## Optically thick lines

$$\tau_{21}(v_r, x, y, z) = \int_{-\infty}^z \kappa_{21}(v_r - v^{(f)}_r, x, y, z') dz', \quad (50)$$

where  $\kappa_{21}$  is the absorption coefficient :

$$\kappa_{21}(v_r, x, y, z) = n_2(x, y, z) \left( \frac{g_2}{g_1} \right) \frac{c^2 A_{21}}{8\pi^{3/2} \nu_{21}^2 \nu_T} e^{(-v_r/\nu_T)^2}, \quad (51)$$

$$I_{21}(v_r, x, y) = \int j_{21}(v_r - v^{(f)}_r, x, y, z) \exp[-\tau_{21}(v_r, x, y, z)] dz. \quad (52)$$

## Summary:

the calculation of the cooling function and the ionisation state of the gas can be done in 3 different degrees of complexity:

- coronal equilibrium or isochoric
- single species, non-equilibrium
- many species, non-equilibrium







